

According to Regulation (EC) No. 1907/2006, Annex II

Methyl Bromide	
8326	
02/01/2012	Revision: 11
14/11/2010	
	8326 02/01/2012

### SECTION 1: Identification of the substance/mixture and of the company/undertaking

1.1 Product identifier	Methyl bromide
Synonym(s)	Bromomethane, MBr
Chemical formula	CH <sub>3</sub> Br
EC No.	200-813-2
<b>REACH Registration No.</b>	The transition time according to REACH Regulation, Article 23 is still not expired
Molecular weight	94.94
Chemical family	Halogenated alkane
CAS number	74-83-9
1.2. Relevant identified uses of the substance or mixture and uses advised against	For industrial use
1.3. Details of the supplier of the Safety Data Sheet	ICL-IP Europe B.V. P.O.Box 465 1000 AL Amsterdam, Netherlands Tel: +31 20 800 5 800 Fax:+31 20 800 5 805 e-mail:msdsinfo@icl-ip.com
Emergency telephone number: - For Europe - For UK and Ireland	( +31) 115 689000 +44 (0) 1270 502891 (24 Hours)



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### **SECTION 2: Hazards identification**

#### 2.1. Classification of the substance or mixture

Classification in accordance with Regulation (EC) No 1272/2008 [CLP/GHS]	<ul> <li>Press. Gas</li> <li>Muta 2, H341 Suspected of causing genetic defects</li> <li>Acute Tox. 3 H301 Toxic if swallowed</li> <li>Acute Tox. 3 H331 Toxic if inhaled</li> <li>STOT RE 2, H373 May cause damage to organs through prolonged or repeated</li> <li>exposure</li> <li>Eye Irrit. 2 H319 Causes serious eye irritation</li> <li>STOT SE 3, H335 May cause respiratory irritation</li> <li>Skin Irrit. 2 H315 Causes skin irritation</li> <li>Aquatic Acute 1, H400 Very toxic to aquatic life</li> <li>Ozone 1 H420, Harms public health and the environment by destroying ozone in the upper atmosphere</li> </ul>
Classification in accordance with Directive 67/548/EEC	<ul> <li>Toxic (T), R 23/25 :Toxic by inhalation and if swallowed.</li> <li>R 68: Possible risk of irreversible effects</li> <li>Xi, R 36/37/38 :Irritating to eyes, respiratory system and skin.</li> <li>Harmful (Xn); R 48/20: Harmful: danger of serious damage to health by prolonged exposure through inhalation</li> <li>Dangerous for the Environment (N), R50: Very toxic to aquatic organisms</li> <li>N, R59 :Dangerous to the ozone layer</li> </ul>

2.2. Label elements

Labelling in accordance with the CLP Regulation EC (No) 1272/2008



Signal word

Danger



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Hazard statements	inhalation. H319 - Causes serious eye H335- May cause respirator H315 - Causes skin irritation H400 - Very toxic to aquatic	to organs through prolonged or repeated exposure irritation ry irritation
Precautionary statements	P260 - Do not breathe fum P280 - Wear protective clo P304 + P340- IF INHALED: comfortable for breathing P305 + P351 + P338 - IF IN Remove contact lenses, if p P302 + P352 - IF ON SKIN P310 - Immediately call a P P330- Rinse mouth.	all safety precautions have been read and understo e/gas/mist/vapours/spray thing/eye protection/face protection. Remove victim to fresh air and keep at rest in a po I EYES: Rinse cautiously with water for several min resent and easy to do. Continue rinsing : Wash with plenty of soap and water. OISON CENTER or doctor/physician

2.3. Other hazards

None



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### **SECTION 3: Composition/information on ingredients**

Components	Weight %	Index No.	EC No.	EU Classification
METHYL BROMIDE	100	# 602-002-00-2	200-813-2	Press. Gas
74-83-9				Muta. 2 H341
				Acute Tox. 3 H301
				Acute Tox. 3 H331
				STOT RE 2 H373
				Eye Irrit. 2 H319
				STOT SE H335
				Skin Irrit 2 H315
				Aquatic Acute 1 H400
				Ozone 1 H420
				(In accordance with CLP
				1272/2008)
				Muta. Cat.3; R68
				N; R50
				N; R59
				T; R23/25
				Xi; R36/37/38
				Xn; R48/20 (In
				accordance with DSD
				67/548/EEC)

### **SECTION 4: First aid measures**

# A 24-HOUR MEDICAL SURVEILLANCE PERIOD IS MANDATORY IN ALL CASES OF EXPOSURE TO METHYL BROMIDE, EVEN IN THE ABSENCE OF ANY IMMEDIATE SIGNS OF POISONING.

#### 4.1. Description of first aid measures

Eye contact	Holding the eyelids apart, flush eyes promptly with copious flowing water for at least 20 minutes. Get medical attention immediately.
Skin contact	Wash skin thoroughly with mild soap and plenty of water for at least 15 minutes. Get medical attention immediately. All leather items should be discarded. Other contaminated clothing must either be discarded or thoroughly ventilated and washed before re-use.



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Inhalation	In case of inhalation, remove person to fresh Keep him quiet and warm. Apply artificial res attention immediately.	
Ingestion	If swallowed, wash mouth thoroughly with pl immediately. NOTE: Never give an unconscious person a	anything to drink.
4.2. Most important symptoms a	nd effects, both acute and delayed	
- Ocular	Severe irritant. Contact with liquid or high co cause severe but usually reversible injury in	
- Dermal	Liquid splashed on clothing or leather or hig skin may cause skin burns with large blisters Less severe exposures may cause itching s May be absorbed through the skin in sufficie	s appearing after several hours. kin rash even after several days.
- Inhalation	Acute poisoning from methyl bromide is cha respiratory tract which may lead, in severe of High concentrations may damage the liver, H Symptoms of poisoning include headache, of vision, slurred speech, nausea and vomiting ONSET OF TOXIC SYMPTOMS MAY BE D SEVERAL DAYS.	ases, to pulmonary edema. kidneys and central nervous system. dizziness, somnolence, vertigo, blurred and possibly convulsions and coma.
- Ingestion	Severe irritant to mucous membranes and to is highly unlikely.	oxic poison if ingested, although ingestion
4.3. Indication of any immediate medical attention and special treatment needed	Intense vesicant. Signs and symptoms of toxicity are primarily and the cardiovascular system. No specific antidote.	referrable to the CNS, respiratory tract

### **SECTION 5: Fire-fighting measures**

**5.1. Extinguishing media** Carbon dioxide, dry chemicals, foam, water spray (fog).



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5.2 Special hazards arising from	I	
the substance or mixture	Although it is considered practically nonflammable, methyl bromide can be ignited with a high energy source of ignition. Containers may rupture violently if exposed to fire or excessive heat for sufficient time. In confined spaces such as buildings or sewers, there is a danger of vapour accumulation, which may result in explosion in the presence of an ignition source. Will decompose from ca. 400°C releasing poisonous and corrosive fumes of carbon monoxide and hydrogen bromide.	
5.3. Advice for fire-fighters	protective clothing. If possible s burning gas unless flow can be	apparatus in positive pressure mode and appropriate stop material flow immediately. Do not extinguish shut off immediately. Use water spray, fog nozzle or nere is no risk, move cylinder away from fire.

### **SECTION 6:** Accidental release measures

6.1. Personal precautions, protective equipment and emergency procedures	Evacuate area and keep personnel upwind. Wear self-contained breathing apparatus in positive pressure mode.
6.3. Methods and materials for containment and cleaning up	If practicable, stop flow of vapour. Ventilate and/or allow to evaporate, keeping people away from the area until safe re- entry levels are shown by halide detector.
6.4 Reference to other sections	None

**SECTION 7: Handling and storage** 

### 7.1. Precautions for safe handling Avoid bodily contact.

Use an appropriate monitoring instrument for methyl bromide in any area where it is
being stored or handled.
Move and transport containers with requisite care. Do not use hooks, rope sling, etc. to unload. Use hand or fork trucks to firmly cradle cylinders.
Do not bump or drag them.

7.2. Conditions for safe storage,	Store containers upright, in a secure manner, either outdoors under ambient
including any incompatibilities	conditions, or indoors in a well ventilated area, away from seeds, foods/feedstuffs
	and human and animal habitation.
	Post as a pesticide storage area. Test periodically for leaks by halide leak detector.



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7.3. Specific end use(s)

Feedstock only

### **SECTION 8: Exposure controls/personal protection**

#### 8.1. Control parameters

Components	Weight %	ACGIH-TLV Data	UK (WEL) - TWA	Germany MAK (TRGS 900) data
METHYL BROMIDE 74-83-9	100	1 ppm skin , A4	5 ppm (20 mg/m <sup>3</sup> ) STEL - 15 ppm (59 mg/m <sup>3</sup> ),10 min	1 ppm (3.9 mg/m³), 3B

#### 8.2. Exposure controls

Ventilation requirements	Ventilation must be sufficient to maintain atmospheric concentration below recommended exposure limit. Mechanical ventilation is recommended. Use local exhaust at source of vapour.
Personal protective equipment: - Respiratory protection	For escape - Gas mask with a new organic vapour canister. For any detectable concentration -
- Hand protection	Self-contained breathing apparatus or supplied-air respirator with a full face-piece. DO NOT WEAR GLOVES when working with MBr because of the danger that liquid or concentrated vapour may be trapped inside them.
- Eye protection	Splash-proof safety glasses. CONTACT LENSES SHOULD NOT BE WORN WHEN WORKING WITH THIS CHEMICAL.
<ul> <li>Skin and body protection</li> </ul>	No specially designed protective clothing is available.
	Do not wear gloves, impervious boots, finger rings or adhesive bandages on hands when handling this material.
Hygiene measures	When using this material, do not eat, drink or smoke. Do not eat, smoke or drink where material is handled, processed or stored. Wash hands carefully before eating or smoking.

### **SECTION 9: Physical and chemical properties**

#### 9.1. Information on basic physical and chemical properties



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SECTION 9: Physical and	d chemical properties
Appearance	Colourless gas, odourless at low concentrations; sweetish odour at very high concentrations. Clear, colourless to straw-coloured liquid under pressure or below 3.5°C.
рН	Not available
Melting point/range	-94°C
Boiling point/range	3.5 - 4°C
Evaporation rate (ether=1)	>1
Vapour pressure	1420 mmHg ( 20°C)
Vapor density	3.3 ( 20°C)
Flash point	None
Flammable/Explosion limits	
- Lower (% vol)	10
- Upper (% vol)	16
Solubility:	
- Solubility in water	0.132 gr/100ml at 25°C (partial pressure CH3Br - 73 torr) 0.138 gr/100ml at 25°C (partial pressure CH3Br - 108 torr)
<ul> <li>Solubility in other solvents</li> </ul>	Infinitely soluble in most organic solvents
Partition coefficient	Log Kow - ~ 1.92
(n-octanol/water)	
Auto-ignition temperature	537°C
Decomposition temperature	~ 400°C
Viscosity	Not applicable
Explosive properties	Not available
Oxidising properties	Not available

### SECTION 10: Stability and reactivity

10.1 Reactivity 10.2 Chemical stability	Decomposes above 400°C Stable in sealed containers and under normal conditions
10.3 Possibility of hazardous reactions	Hazardous polymerisation will not occur
10.4 Conditions to avoid	Keep away from ignition sources Avoid contamination by water
10.5 Incompatible materials	Strong oxidizers, aluminum, tin, zinc and magnesium metals and their alloys, natural rubber and certain types of plastics.
10.6 Hazardous decomposition products	CO, HBr



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### **SECTION 11: Toxicological information**

11.1 Information on toxicologie	cal effects	
Acute toxicity: - Rat oral LD50	liquid MBr in corn oil - 104 mg/kg microencapsulated MBr in corn oil - 133 mg/kg	
- Rat inhalation LC50	1175 mg/m <sup>3</sup> /8 hour	
- Mouse inhalation LC50 Serious eye damage/ irritation	1540 mg/m³/2 hour Severe irritant	
Skin corrosion/irritation	Irritant	
Respiratory or skin sensitisation	Exposure in human resulted in redness, congestion, dermatitis, itching, swollen areas and blistering.	
Mutagenicity	Mutagenic by the Ames Test MBr induced DNA damage in rat testis following inhalation exposure at 250 ppm (6 hours/day for 5 consecutive days). In vivo, MBr induced sister chromatid exchanges in bone marrow cells and micronuclei in peripheral erythrocytes of female mice exposed by inhalation for 14 days.	
Carcinogenicity	Studies conducted with MBr, exposing animals both by inhalation (rats & mice) and by oral route (fumigated feed, rats), showed that THERE WAS NO EVIDENCE OF CARCINOGENIC ACTIVITY. Not included in NTP 12th Report on Carcinogens IARC Group 3 (animal inadequate evidence, human no data available)	
Reproductive toxicity	In a two generation reproductive study via inhalation in albino rats, the NOEL was 90	
Specific Target Organ Toxicity (STOT) - Single exposure	ppm. May cause respiratory irritation	
Specific Target Organ Toxicity (STOT) - Repeat exposure	Chronic exposure to low concentrations of methyl bromide may produce central nervous system effects. Signs include mental confusion, lethargy, inability to focus one's eye, incoordination and muscle weakness. Repeated skin contact may cause dermatitis.	
Aspiration hazard	Not expected to occur	



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Other	Single exposure vapour inhalation neurotoxicity study in rats: NOEL - 100 ppm Acute oral toxicity (single dose) study in Beagle dogs: Lethal dose - 500 mg/kg No clinical signs were observed at 1 mg/kg	

### **SECTION 12: Ecological information**

12.1 Toxicity Aquatic toxicity : - 96 Hour-LC50, Fish	3.9 mg/l (Rainbow Trout) 56.28 mg/l (Zebrafish)
<ul> <li>- 48 Hour-EC50, Daphnia magna</li> <li>- 72 Hour-EC50, Freshwater alga</li> <li>Avian toxicity:</li> </ul>	
- Oral LD50 12.2 Persistence and degradability	~ 73 mg/kg (Northern Bobwhite) 12.2 Persistence and degradability
- Hydrolysis	Under laboratory conditions (MBr) Half-life at pH 5 - 256.7 hours Half-life at pH 7 - 253.9 hours Half-life at pH 9 - 357.3 hours
12.3 Bioaccumulative potential	Not bioaccumulative
12.4 Mobility in soil	There is no accumulation of methyl bromide per se, since it is rapidly degraded in the soil.
12.5 Results of PBT and vPvB assessment	Not considered to be PBT or vPvB
12.6 Other adverse effects	Germany, water endangering classes (WGK) 3 Methyl bromide is listed in the Montreal Protocol as a controlled substance with an ODP (Ozone Depleting Potential) of 0.6.



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SECTION 13: Disposal co	onsiderations	
13.1 Waste treatment methods		
Waste disposal	The recommended method is incineration. chamber is not available, return MARKED Contact local and/or state environmental a Observe all federal, state and local enviror material.	containers to supplier. uthorities to insure proper compliance.
SECTION 14: Transport	information	
UN No.	1062	
ADR/RID	Proper shipping name: Methyl bromide Hazard identification No. 26 Class 2 : Gases Classification Code: 2T Label No.: 2.3+13(RID) Marking: Environmentally hazardous subs	tance
IMO	Proper shipping name: Methyl bromide	

**SECTION 15: Regulatory information** 

#### 15.1 Safety, health and environmental regulations/legislation specific for the substance or mixture

Proper shipping name: Methyl bromide

Marking: Environmentally hazardous substance

Class: 2.3 Toxic Gases Label: TOXIC GAS (2) Mark: MARINE POLLUTANT

Cargo aircraft - Forbidden Passenger aircraft - Forbidden

Class: 2.3

EU

ICAO/IATA

Regulated under Article 22 of EC Regulation No. 2037/2000 on substances that deplete the ozone layer.



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USA	Reported in the EPA TSCA Inventory. This p	product is subject to registration under
Australia	Listed in AICS	
Canada China inventory	Listed in DSL This substance is listed under Part 1, Group Release Inventory (NPRI) for 2008. Informat reported to the Minister of the Environment i the Canadian Environmental Protection Act, This chemical is included on the current pha substances under the Canadian Environment Listed in IECSC	tion about this substance must be n accordance with subsection 46(1) of 1999. Ise-out schedule of ozone-depleting
Japan	ENCS no. 2-39 ISHL no. 2-39	
Korea	Listed in ECL (KE-03676) Toxic chemical No.97-1-113, 1% or more in	mixtures (MBr)
Philippines	Listed in PICCS	
Hong Kong	Dangerous Goods - Category 2 - Compress Ozone Depleting Substances - Part 6 sched	
Taiwan	Harmful substances	
15.2 Chemical Safety Assessme	ntA Chemical Safety Assessment has not yet Regulation.	been carried out under the REACH

**SECTION 16: Other information** 

This data sheet contains changes from the previous version in section(s) 2, 3, 8 (REACH format) 14



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#### Health, Safety & Environment Policy

We will strive to ensure that our operations and products meet the needs of the present global community without compromising the ability of future generations to meet their needs

We accept that the success of our business is dependent on the supply of products and services that will benefit society whilst ensuring human safety and protection of the environment and natural resources

Within the framework of our commitment to the Responsible Care program, we will provide a healthy and safe work environment for employees and will responsibly manage our products at all stages of their life cycle in order to protect human health and the environment whilst maintaining high production standards of operation

TO MEET THIS COMMITMENT WE WILL: Comply with or exceed applicable national and international regulatory requirements and other requirements to which we subscribe Communicate openly and actively encourage dialogue with employees, customers and community concerning our products and operations Implement documented management systems consistent with and for promotion of the Responsible Care ethics

Develop and supply products that can be manufactured, transported, used and disposed of safely whilst best meeting the needs of our customers Regularly assess, continually improve and responsibly manage health, safety and environmental risks associated with products and processes throughout their life-cycles Share knowledge and expertise with others and seek to learn from and incorporate improved practices into our own operations

Educate and train employees, contractors and customers to improve their HSE performance Communicate up-to-date information to enable our workers, customers and other interested parties to handle our products in a safe and environmentally responsible manner Endeavor to work with customers, suppliers, distributors and contractors to foster the safe use, transport and disposal of our chemicals Support Product Stewardship programs in cooperation with customers, distributors and transporters

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End of safety data sheet